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- 3 The table below shows the nucleon number, the number of neutrons and number of electrons in particles X, Y, Z, Q, R, T and U.  
 Jadual di bawah menunjukkan nombor nukleon, bilangan neutron dan bilangan elektron bagi zarah X, Y, Z, Q, R, T dan U.

Particles / Zarah	X	Y	Z	Q	R	T	U
Nucleon number / Nombor nukleon	20	24	23	16	12	27	35
Number of proton / Bilangan proton	10	12	11	8	6	13	17
Number of neutron / Bilangan neutron	10	12	12	8	6	14	18
Number of electron / Bilangan elektron	10	10	11	10	6	10	17

- (a) What is meant by nucleon number?  
 Apakah maksud nombor nukleon?

The total number of proton and neutron in the nucleus of an atom.

Jumlah proton dan neutron dalam satu atom.

- (b) Complete the number of proton of the particles in the table above.  
 Lengkapkan bilangan proton bagi zarah dalam jadual di atas.

- (c) State a particle which is  
 Nyatakan zarah yang merupakan

- (i) an atom of a non-metal / atom bukan logam  
 (ii) an atom of a metal / atom logam  
 (iii) a positive ion / ion positif  
 (iv) a negative ion / ion negatif  
 (v) a positive ion with charge 3+ / ion positif dengan cas 3+  
 (vi) an atom of a noble gas / atom gas adi

X / R / U

Z

Y / T

Q

T

X

- (d) Particle Y combines with particle Q to form a compound.  
 Zarah Y bergabung dengan zarah Q untuk membentuk sebatian.

- (i) State the type of compound formed.  
 Nyatakan jenis sebatian yang terbentuk.

Ionic compound

Sebatian ion

- (ii) Write chemical formula for the compound formed.  
 Tuliskan formula kimia bagi sebatian yang terbentuk.

YQ

- (iii) Draw the electron arrangement of the compound formed.  
 Lukiskan susunan elektron bagi sebatian yang terbentuk.



- (e) Particle R combines with particle U to form a compound.  
 Zarah R bergabung dengan zarah U untuk menghasilkan suatu sebatian.

- (i) State the type of compound formed.  
 Nyatakan jenis sebatian yang terbentuk.

Covalent compound

Sebatian kovalen.

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